Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Peter Kirsop, John M. D. Storey and William T. A. Harrison*

Department of Chemistry, University of Aberdeen, Meston Walk, Aberdeen, Scotland

Correspondence e-mail: w.harrison@abdn.ac.uk

Key indicators

Single-crystal X-ray study T = 120 K Mean σ (C–C) = 0.009 Å Disorder in main residue R factor = 0.055 wR factor = 0.133 Data-to-parameter ratio = 18.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

4-(3-Allyloxy-2-bromophenoxy)but-2-enenitrile

The title compound, $C_{13}H_{12}BrNO_2$, possesses normal geometrical parameters. The molecular conformation and crystal packing appear to be influenced by intramolecular $C-H\cdots O$ and intermolecular $C-H\cdots N$ interactions, the latter resulting in centrosymmetric $R_2^2(10)$ loops.

Received 16 August 2004 Accepted 20 August 2004 Online 28 August 2004

Comment

The title compound, 4-(3-allyloxy-2-bromophenoxy)but-2enenitrile, (I) (Fig. 1), arose during our studies to determine the philicity of aryl radicals by competetive cyclization (Kirsop *et al.*, 2004a,b,c).



Compound (I) possesses normal geometrical parameters (Table 1). The C12–C13 bond was flagged (Spek, 2003) as being longer than expected for a Csp^2-Csp bond, but similar equivalent C–C bond lengths have been seen in other cyanoethenyl groupings [*e.g.* Ishii *et al.* (2000); bond length = 1.424 (6) Å]. The *exo* C1–C2 vinyl group in (I) is disordered over two positions, with refined relative occupancies of 0.710 (19):0.210 (19) for the isotropically refined major (C1*a*–C2*a*) and minor (C1*b*–C2*b*) components, respectively. The C4–C9 phenyl ring and its attached non-H atoms (O1, O2 and Br1) is statistically flat [r.m.s. deviation from the best least-squares plane = 0.011 Å, maximum deviation = 0.018 (3) Å for O2]. A *PLATON* (Spek, 2003) analysis of (I) indicated that the backbone of the nitrile side chain may be stabilized by an



Figure 1

View of (I) (50% displacement ellipsoids for the non-H atoms). Only the major disorder component for C1 and C2 is shown and H atoms are drawn as small spheres of arbitrary radius. The possible intramolecular $C-H \cdots O$ interaction is indicated by a dashed line.

01636 Peter Kirsop et al. \cdot C₁₃H₁₂BrNO₂

© 2004 International Union of Crystallography

Printed in Great Britain - all rights reserved

intramolecular C-H···O interaction (Fig. 1 and Table 2), thereby helping to establish an essentially planar arrangement of atoms O2, C10, C11 and C12 (Table 1). Dimers of (I) associate together by way of a pair of inversion-symmetrygenerated C-H···N interactions (Table 2 and Fig. 2), resulting in centrosymmetric $R_2^2(10)$ loops (Bernstein *et al.*, 1995). There are no significant $\pi - \pi$ interactions in the unit cell of (I); the unit-cell packing is shown in Fig. 3.

Experimental

Crotonitrile (2.00 g, 0.03 mol), *N*-bromosuccinamide (6.3 g, 0.04 mol) and azoisobutyronitrile (0.49 g, 0.003 mol) were added to dry acetone (100 ml). The mixture was stirred at reflux under a nitrogen atmosphere for 4 h. After cooling, the solvent was removed at reduced pressure to give a quantity of dark-brown oil. Distillation under reduced pressure yielded bromocrotonitrile as a pale-brown oil (b.p. 345–347 K at 7 mm Hg) (3.5 g, 80%). ¹H NMR: δ_H (CDCl₃) 4.1 (2H, d, J = 7.0 Hz), 5.5 (1H, d, J = 12.5 Hz), 6.7 (1H, m).

3-Allyloxy-2-bromophenol (2.00 g, 0.009 mol) (Kirsop et al., 2004c), bromocrotonitrile (1.58 g, 0.011 mol) and potassium carbonate (8.00 g, 0.058 mol) were added to dry acetone (100 ml). The mixture was stirred at reflux under a nitrogen atmosphere for 3 h. After cooling, the mixture was filtered and the solvent removed at reduced pressure to give a pale-yellow oil (1.81 g, 68%). Thin-layer chromatography (4:1 hexane-ethyl acetate) showed 4-(3-allyloxy-2bromophenoxy)but-2-enenitrile, (I), as a sharp spot at $R_F = 0.38$. The crude product was purified by flash column chromatography to give (I) as a white powder (1.42 g, 54%). A sample of this powder was recrystalized from hot hexane-ethyl acetate (20:1) to give white needles (m.p. 349–351 K). ¹H NMR: δ_H (CDCl₃) 4.58 (2H, m, CH₂), 4.68 (2H, m, CH₂), 5.27 (1H, d, J = 10.6 Hz, CH), 5.45 (1H, d, J = 17.1 Hz, CH), 5.97 (1H, m, CH), 6.03 (1H, m, CH), 6.45 (1H, d, J = 8.5 Hz, Ar-H), 6.59 (1H, d, J = 8.5 Hz, Ar-H), 6.81 (1H, dt, J = 16.0 and 3.5 Hz, CH), 7.16 (1H, t, J = 8.5 Hz, Ar-H). ¹³C NMR: $\delta_{\rm C}$ 67.1, 69.8, 101.0, 102.4, 105.9, 107.2, 116.9, 117.8, 128.3, 132.4, 147.8, 155.2, 156.5. $\nu_{\rm max}$ (KBr)/cm⁻¹ 2226, 1588, 1471, 1258, 1120, 1052, 759.

Crystal data

| $\begin{array}{l} C_{13}H_{12}\text{BrNO}_2 \\ M_r = 294.15 \\ \text{Monoclinic, } P2_1/c \\ a = 4.3681 (3) \text{ Å} \\ b = 19.1969 (12) \text{ Å} \\ c = 15.0270 (9) \text{ Å} \\ \beta = 97.570 (4)^{\circ} \\ V = 1249.09 (14) \text{ Å}^3 \\ Z = 4 \end{array}$ | $D_x = 1.564 \text{ Mg m}^{-3}$ Mo K\alpha radiation Cell parameters from 2721 reflections $\theta = 2.9-27.5^{\circ}$ $\mu = 3.28 \text{ mm}^{-1}$ T = 120 (2) K Needle, colourless $0.40 \times 0.03 \times 0.02 \text{ mm}$ |
|--|---|
| Data collection Nonius KappaCCD diffractometer φ and ω scans Absorption correction: multi-scan (SORTAV; Blessing, 1995) $T_{min} = 0.354, T_{max} = 0.937$ 17155 measured reflections 2871 independent reflections | 1930 reflections with $I > 2\sigma(I)$ $R_{int} = 0.089$ $\theta_{max} = 27.6^{\circ}$ $h = -5 \rightarrow 5$ $k = -24 \rightarrow 24$ $l = -19 \rightarrow 19$ |
| Refinement | |

Refinement on F^2

| Kennement on r |
|---------------------------------|
| $R[F^2 > 2\sigma(F^2)] = 0.055$ |
| $wR(F^2) = 0.133$ |
| S = 1.02 |
| 2871 reflections |
| 153 parameters |
| H-atom parameters constrained |

$$\begin{split} w &= 1/[\sigma^2(F_o^2) + (0.0553P)^2 \\ &+ 1.5353P] \\ \text{where } P &= (F_o^2 + 2F_c^2)/3 \\ (\Delta/\sigma)_{\text{max}} < 0.001 \\ \Delta\rho_{\text{max}} = 0.46 \text{ e } \text{\AA}^{-3} \\ \Delta\rho_{\text{min}} &= -0.58 \text{ e } \text{\AA}^{-3} \end{split}$$



Figure 2

A dimeric association of two molecules of (I) via two C-H···N interactions (dashed lines). The symmetry code is as in Table 2. All H atoms except for H11 and H12 have been omitted for clarity.



Figure 3

Unit-cell packing in (I), viewed down [100] (50% displacement ellipsoids). All H atoms except for H11 have been omitted for clarity.

Table 1

Selected geometric parameters (Å, °).

| C9-Br1 | 1.888 (4) | C12-C13 | 1.437 (6) |
|----------------|-----------|-----------------|-----------|
| C11-C12 | 1.318 (6) | C13-N1 | 1.148 (5) |
| C1a-C2a-C3-O1 | 111.3 (7) | C10-C11-C12-C13 | 179.0 (4) |
| O2-C10-C11-C12 | -4.0(6) | | |

Table 2

Hydrogen-bonding geometry (Å, °).

| , , , | 50 50 | <i>, ,</i> | | |
|------------------------------------|------------|-------------------------|--------------|------------------|
| $D - H \cdot \cdot \cdot A$ | D-H | $H \cdot \cdot \cdot A$ | $D \cdots A$ | $D - H \cdots A$ |
| C12-H12···O2 | 0.95 | 2.31 | 2.671 (5) | 102 |
| $C11 - H11 \cdot \cdot \cdot N1^i$ | 0.95 | 2.51 | 3.410 (6) | 158 |
| Symmetry code: (i) 2 | -x - y - z | | | |

Symmetry code: (i) 2 - x, -y, -z.

All the H atoms were placed in idealized locations and refined by riding on their carrier atoms (C–H = 0.95–0.99 Å). For all H atoms, the constraint $U_{iso}(H) = 1.2U_{eq}(\text{parent atom})$ was applied.

Data collection: *COLLECT* (Nonius, 1998); cell refinement: *HKL SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *HKL DENZO* (Otwinowski & Minor, 1997) and *SCALEPACK* and *SORTAV* (Blessing, 1995); program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *ORTEP3* (Farrugia, 1997); software used to prepare material for publication: *SHELXL*97.

We thank the EPSRC UK National Crystallography Service (University of Southampton) for the data collection.

References

 Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). Angew. Chem. Int. Ed. Engl. 34, 1555–1573.
 Blessing, R. H. (1995). Acta Cryst. A51, 33–38.

Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.

- Ishii, K., Shimada, Y., Sugiyama, S. & Noji, M. (2000). J. Chem. Soc. Perkin Trans. 1, pp. 3022–3024.
- Kirsop, P., Storey, J. M. D. & Harrison, W. T. A. (2004a). Acta Cryst. E60, 0222-0224.
- Kirsop, P., Storey, J. M. D. & Harrison, W. T. A. (2004b). Acta Cryst. C60, 0353-0355.
- Kirsop, P., Storey, J. M. D. & Harrison, W. T. A. (2004c). Acta Cryst. E60, 01147–01148.
- Nonius (1998). COLLECT. Program for data collection. Nonius BV, Delft, The Netherlands.
- Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography*, Part A, edited by C. W. Carter Jr and R. M. Sweet, pp. 307–326. New York: Academic Press.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.